## **ME 301: Fundamentals of Thermodynamics**

This is a cooperative course taught jointly by WSU and the University of Idaho

Course description:	Thermodynamic properties of matter, ideal gas, real fluids, and incompressible substances, work and heat, first and second laws and application to engineering systems		
Number of credits:	3. This course is required.		
Course Coordinator:	J. Leachman		
Prerequisites by course:	Phys 201 with a grade of C or better, or PHYS 201 and 211, both with C or better.		
Prerequisites by topic:	<ol> <li>Differentiation</li> <li>Integration</li> <li>Conservation of Mass</li> <li>Conservation of Energy</li> </ol>		
Postrequisites:	ME 405; ME 303 (recommended)		
Textbooks/other required materials:	<ol> <li>Cengel, Y. Property Tables Booklet/Thermodynamics. McGraw-Hill, 2024, 10/e.</li> <li>Cengel, Y.A. and Boles, M.A. Thermodynamics: An Engineering Approach. McGraw-Hill, 2024, 10/e.</li> </ol>		
Course objectives:	<ol> <li>Calculate changes in internal energy and enthalpy using real fluid, ideal gas, and incompressible property models.</li> <li>Apply the first law of thermodynamics to closed and open systems.</li> <li>Calculate changes in entropy using real fluid, ideal gas, and incompressible property models.</li> <li>Apply the second law of thermodynamics to closed and open systems.</li> <li>Analyze Carnot, power, and refrigeration cycles using state-diagrams.</li> </ol>		
Topics covered:	<ol> <li>Basic concepts of properties in pure substance.</li> <li>First law of thermodynamics for closed systems.</li> <li>First law of thermodynamics for control volumes.</li> <li>Second law of thermodynamics; Carnot Cycle; thermodynamic temperature scale.</li> <li>Concept and calculation of entropy.</li> <li>Gas power cycles; Vapor cycles.</li> <li>Refrigeration cycles.</li> </ol>		
Expected student outcomes:	<ol> <li>An understanding of how an automobile engine runs, how a utility plant generates electricity, and how a refrigerator keeps the icebox cold.</li> <li>Ability to analyze the performance of an engine, a power plant, or a refrigerator by applying the first law of thermodynamics.</li> <li>Ability to determine the fundamental limits on the operation of these devices using the second law of thermodynamics.</li> </ol>		
Class schedule: Laboratory schedule: Contribution to meeting the professional component:	Three 50-minute lecture sessions per week, for one semester. None Engineering Topics		
Relationship to student outcomes:	Strongly Supports Outcome 1		
Prepared by:	Chris Doman and J. Leachman Date: January 3, 2023		

A. **Reasonable Accommodation:** Reasonable accommodations are available for students with a documented disability. If you have a disability and need accommodations to fully participate in this class, please either visit or call the Access Center (Washington Building 217; 509-335-3417) to schedule an appointment with an Access Advisor. All accommodations MUST be approved through the Access Center.

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- B. WSU Safety: WSU is committed to maintaining a safe environment for its faculty, staff, and students. Safety is the responsibility of every member of the campus community and individuals should know the appropriate actions to take when an emergency arises. In support of our commitment to the safety of the campus community the University has developed a <u>Campus Safety Plan (http://safetyplan.wsu.edu)</u>. It is highly recommended that you visit this web site as well as the University emergency management web site at <u>http://oem.wsu.edu/</u> to become familiar with the information provided.
- C. Academic Integrity: Academic Integrity is the cornerstone of higher education. As such, all members of the University community share responsibility for maintaining and promoting the principles of integrity in all activities, including academic integrity and honest scholarship. Academic integrity will be strongly enforced in this course. Students who violate WSU's Academic Integrity Policy (identified in Washington Administrative Code (WAC) 504-26-010(4) will receive [insert academic sanction (e.g., fail the course, fail the assignment, etc.)], will not have the option to withdraw from the course pending an appeal, and will be reported to the Center for Community Standards. Cheating includes, but is not limited to, plagiarism and unauthorized collaboration as defined in the Standards of Conduct for Students, WAC 504-26-010(3). You need to read and understand all of the definitions of cheating. If you have any questions about what is and is not allowed in this course, you should ask course instructors before proceeding.

#### **Specifics for Spring 2024 Section 1**

# MEETING TIME AND LOCATION:

Lecture: Goertzen 21, MWF 2:10-3:00.

**<u>Recitation</u>**: Spark 339, F 3:10-5:00. \*\*Note that recitation functions as office hours for this class and is intended as a group work time. If you have a private matter to discuss with me, please send me an email or call my cellphone.

**INSTRUCTOR:** Dr. Jacob Leachman, Office: HYPER Lab 230 Dairy Road, Phone: 208-816-0288 (cell) e-mail: <u>jacob.leachman@wsu.edu</u> \*\* I'm usually in meetings or in my office across campus. Please call my cellphone if it's urgent.

TEACHING ASSISTANT(s): Parsa Akbari, Office: ETRL 207, Phone: 509-338-8384, parsa.akbari@wsu.edu.

#### ADDITIONAL COURSE RESOURCES:

- ✓ <u>Alternative Text</u>: S.A. Klein and G.F. Nellis, *Thermodynamics*, Cambridge University Press, 1<sup>st</sup> ed. (2011). Also can be viewed as an e-book via the WSU library.
- $\checkmark$  <u>Website</u>: All course materials will be posted on Canvas.

#### **HOMEWORK:**

- Homework is due every Friday at 8 pm.
- No credit is awarded for late assignments; your two lowest homework grades for the class will be replaced with perfect scores. Extenuating circumstances will be evaluated on a case-by-case basis when I am notified in advance.
- Given the large volume of homework that will be graded, it is in your best interest to make your work as neat as possible, credit may be deducted for solutions that are not clearly identified and/or difficult to follow.
- Students must complete all parts of the homework on their own; however, I encourage you to work with friends to teach each other solution strategies as this still allows for differentiation of original solutions. Copying, pasting, reusing code, or handing in the same work with different names will result in immediate submission to the student conduct board, potentially without warning.

#### EXAMS:

- There will be three, class-long, cumulative examinations.
- The TA and myself will take attendance and proctor the exams. You are allowed to use a SCIENTIFIC calculator as these are required for use on the Fundamentals of Engineering (FE) exam.

**GRADING:** Homework: 50%; 3 Exams: 50%. <u>Distribution</u>: 100–93 (A), 93–90 (A-), 90–87 (B+), 87–83 (B), 83–80 (B-), 80–77 (C+), 77–73 (C), 73–70 (C-), 70–60 (D), 60–0 (F), incomplete (I), integrity violation (X).

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### **SCHEDULE:**

\*the class schedule topics may change during the semester

Date		Day	Topics	Readings		
1/8		•	1. Class Organization, Introduction to Thermodynamics	0		
1/10		Wed	· · · · · · · · · · · · · · · · · · ·	Chapter 1		
1/12	s	Fri	3. The forms of energy, heat, and work	Chapter 2		
1/15	nic	Mon				
1/17	าลท	Wed	· · ·	Chapter 3		
1/19	dyı	Fri	5. Properties of pure substances: Real fluids	Chapter 3		
1/22	no	Mon	6. Closed system energy balance and internal energy	Chapter 4		
1/24	Thermodynamics	Wed		Chapter 4		
1/26	Th	Fri	8. Open system energy balance and enthalpy	Chapter 4,5		
1/29	of	Mon	9. Open un-steady system energy balance	Chapter 5		
1/31	аw	Wed		Chapter 5		
2/2	1 <sup>st</sup> Law	Fri	11. Analysis of open steady systems: Pumps, Compressors, Diffusers	Chapter 5		
2/5	$1^{s_1}$	Mon	12. Analysis of open steady systems: Nozzles, Throttles, Turbines	Chapter 5		
2/7		Wed	13. 1 <sup>st</sup> Law Demo Day	Chapter 5		
2/9		Fri	Exam #1			
2/12			14. The concept of Irreversibility and the 2 <sup>nd</sup> Law	Chapter 6		
2/14		Wed		Chapter 6		
2/16		Fri	16. Entropy and Thermodynamic Temperature Scales	Chapter 6		
2/19	ics	Mon	President's Day – CLASS HOLIDAY			
2/21	of Thermodynamics	Wed		Chapter 7		
2/23	yn:	Fri	18. Closed system entropy balance: Incompressible and Ideal	Chapter 7		
2/26	pol	Mon	19. Open system entropy balance	Chapter 7		
2/28	um	Wed	20. Open, un-steady system entropy balance	Chapter 7		
3/1	Che	Fri	21. Isentropic efficiency: Pumps, Compressors, Diffusers	Chapter 8		
3/4	f 1	Mon	22. Isentropic efficiency: Nozzles, Throttles, Turbines	Chapter 8		
3/6	w (	Wed		•		
3/8	Law	Fri	24. Availability/Exergy analysis of closed & open systems			
	$2^{\rm nd}$		Spring Vacation			
3/18		Mon	25. Entropy optimization of flow systems			
3/20		Wed				
3/22		Fri	Exam #2			
3/25		Mon	27. Power Cycles: Concepts and Carnot	Chapters 9,10		
3/27		Wed		Chapters 9,10		
3/29		Fri	29. Power Cycles: Rankine cycle modifications	Chapters 9,10		
4/1	20	Mon	30. Power Cycles: Brayton for power	Chapters 9,10		
4/3	Cycles		31. Power Cycles: Brayton for propulsion	Chapters 9,10		
4/5	Cy	Fri	32. Power Cycles: Otto & Diesel cycles	Chapters 9,10		
4/8	nic	Mon	33. Power Cycles: Stirling cycles	Chapters 9,10		
4/10	am	Wed	34. Refrigeration Cycles: Concepts and Carnot	Chapter 11		
4/12	Thermodynamic	Fri	35. Refrigeration Cycles: Standard vapor compression	Chapter 11		
4/15	Jod	Mon	36. Refrigeration Cycles: Heat pump systems	Chapter 11		
4/17	ern	Wed	37. Refrigeration Cycles: Advanced and combination cycles	Chapter 11		
4/19	Γhε	Fri	38. Refrigeration Cycles: Refrigerant properties and selection	Chapter 11		
4/22	-	Mon	39. Cycles Review			
4/24		Wed	40. Thermodynamics and cryogenic hydrogen research			
4/26		Fri	41. FE Exam Prep			
5/1		Wed	Exam #3 1:30-3:30 pm			